

DE LA RECHERCHE À L'INDUSTRIE

cea



Molecular science for artificial photosynthesis

From bio-inspired catalyst to nanomaterials

V. Artero

*Laboratoire de Chimie et Biologie des Métaux,
Université Joseph Fourier, CNRS, CEA Grenoble*
www.solhycat.com



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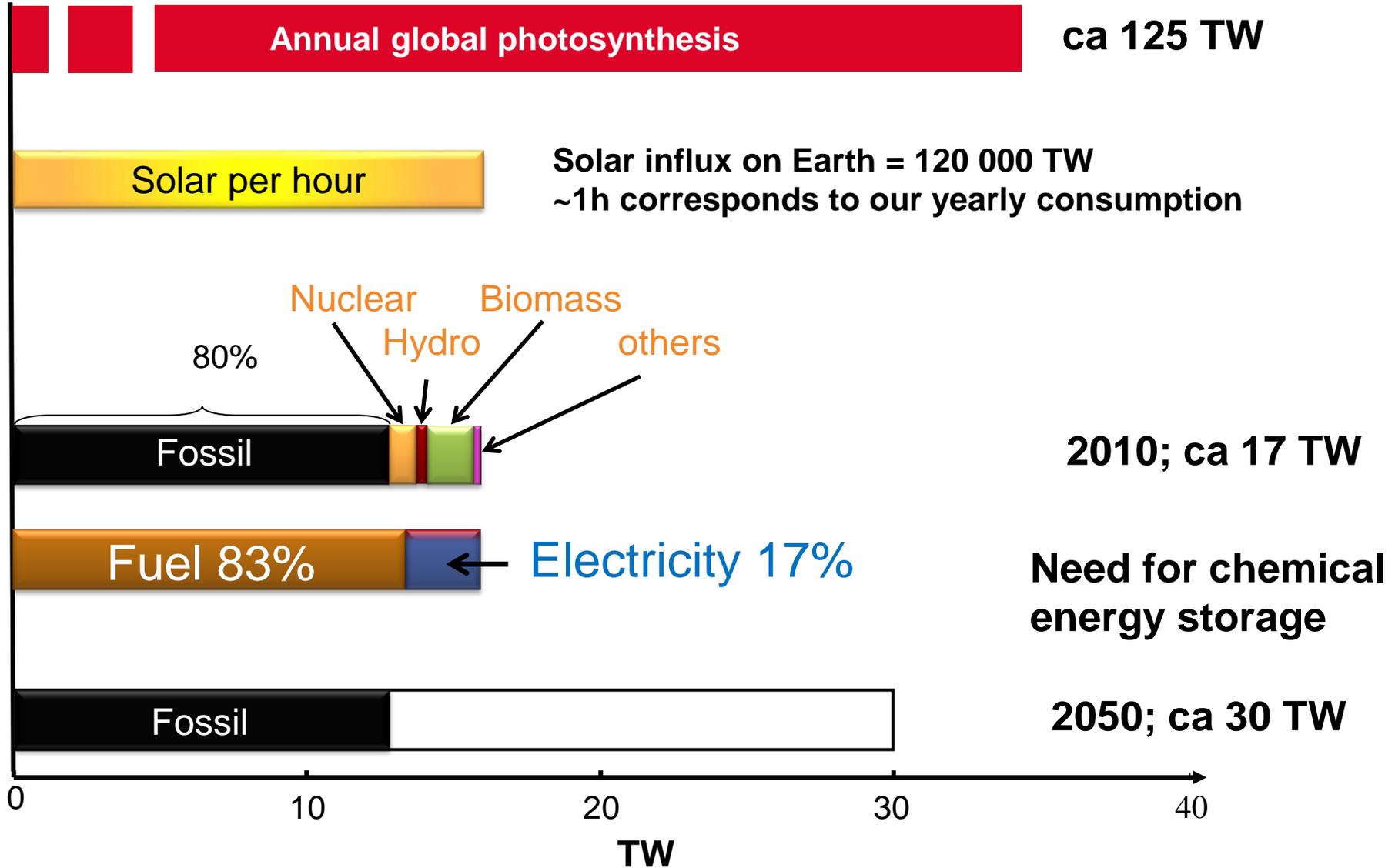
IPCM- Univ. Paris 6

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LCMCP- Univ. Paris 6

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The global energy challenge



Renewable energy: Solar

Renewable energy



- Concentration
- Storage (fuels)

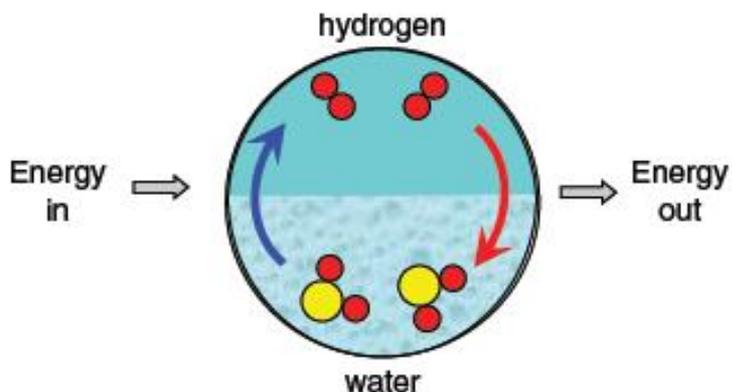


Hydrogen (H₂)
production

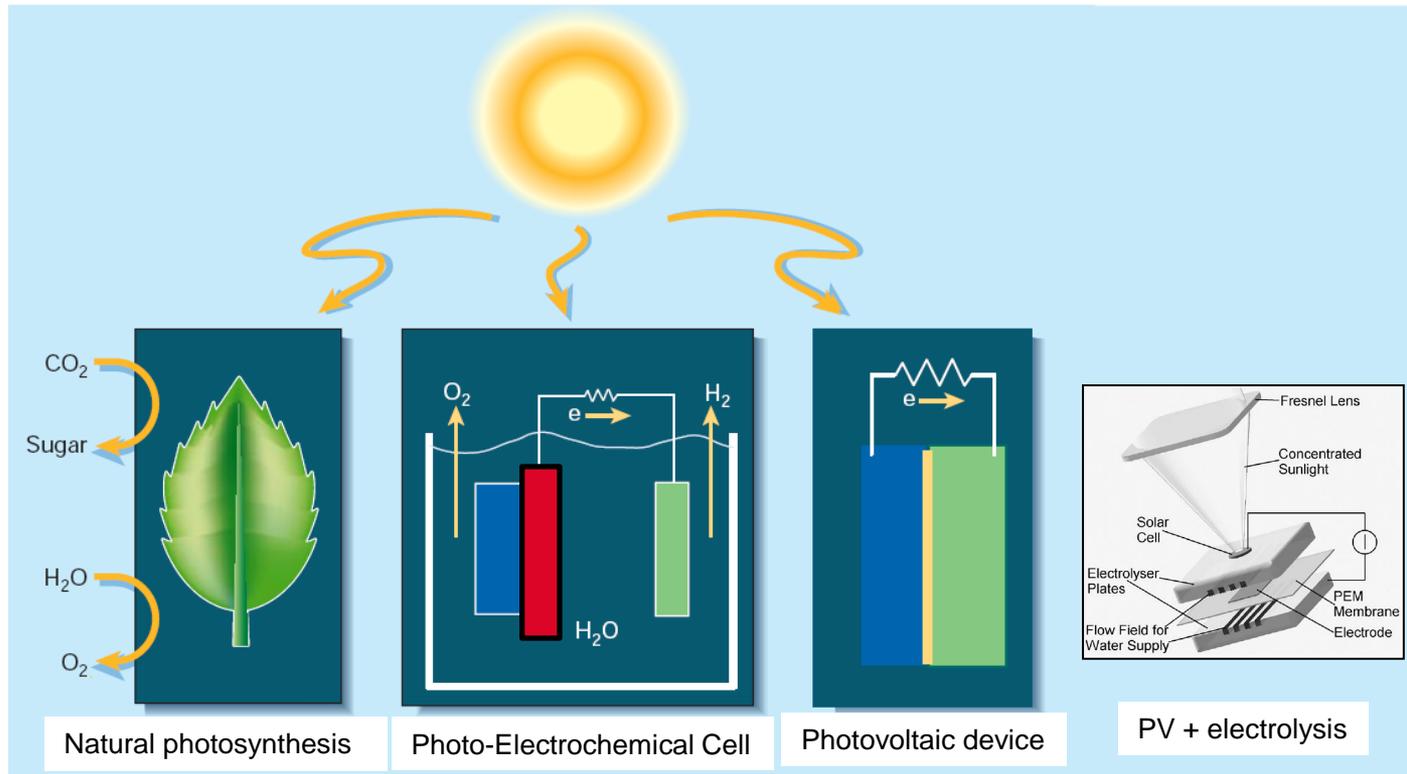


CO₂ recycling and
production of liquid fuels

Li-ion batteries:	0.46 - 0.72 MJ.kg ⁻¹
Oil:	47 MJ.kg ⁻¹
Hydrogen:	140 MJ.kg ⁻¹



Converting solar energy into chemical energy



State of the art « PV + Electrolysis » technology

PV + electrolysis

Wired PV + Electrolysis

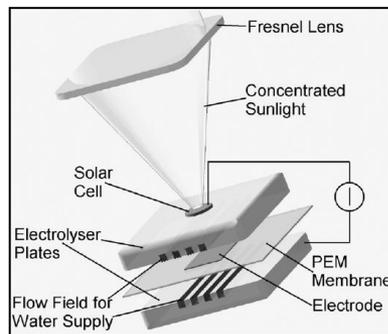
👍 👎 Robust- Mature (>18%)

👎 3 devices

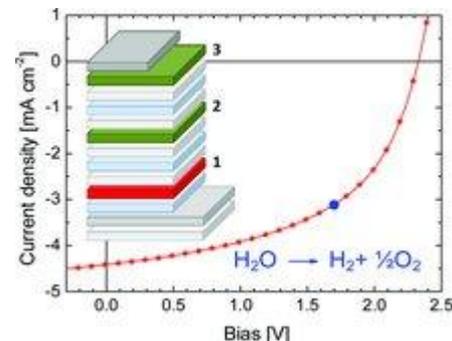
PV

Electrolysis cell

Power management



Peharz IJHE 2007



Janssen Adv. Mater. 2013

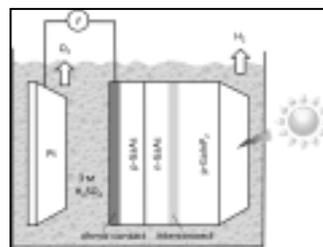
Unwired PV + Electrolysis

Energy & Fuels 1998, 12, 3–10

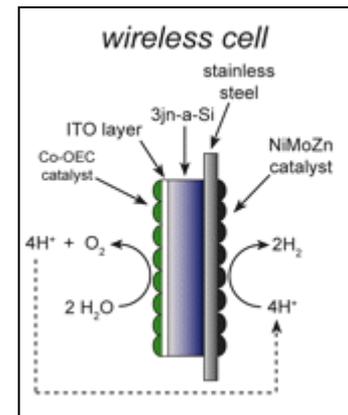
High-Efficiency Photoelectrochemical Hydrogen Production Using Multijunction Amorphous Silicon Photoelectrodes

Richard E. Rocheleau,* Eric L. Miller, and Anupam Misra

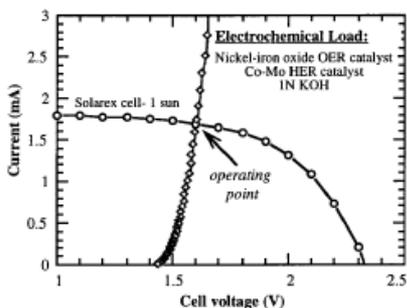
Hawaii Natural Energy Institute, School of Ocean and Earth Science and Technology, University of Hawaii at Manoa, Honolulu, Hawaii 96822



Turner Science 1998



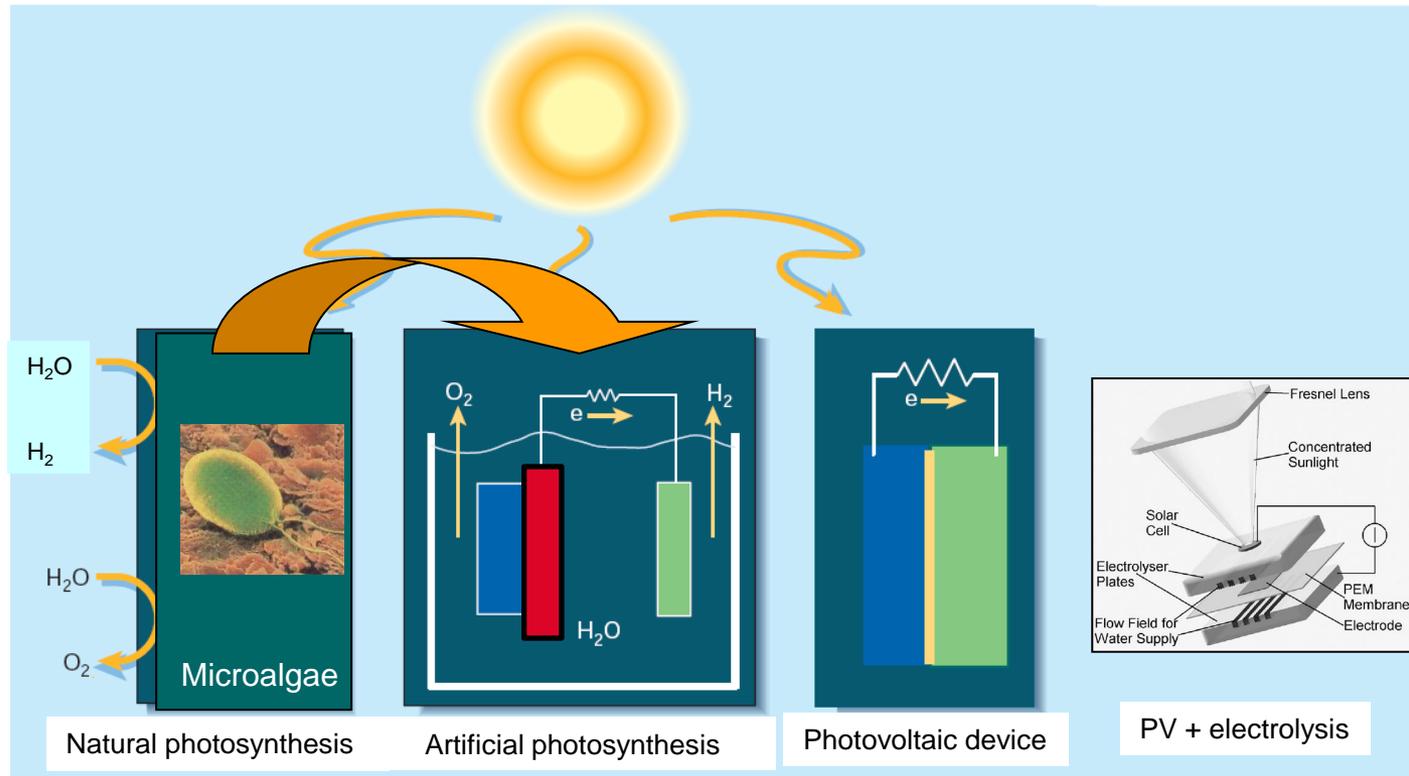
Nocera Science 2011



$\eta = 7.8\%$
7200 h

- 👍 Single device
- 👎 Stability of PV materials into electrolyte
- 👎 One fixed operating point

Converting solar energy into chemical energy



Abundant and cheap materials both
for light harvesting and catalysis

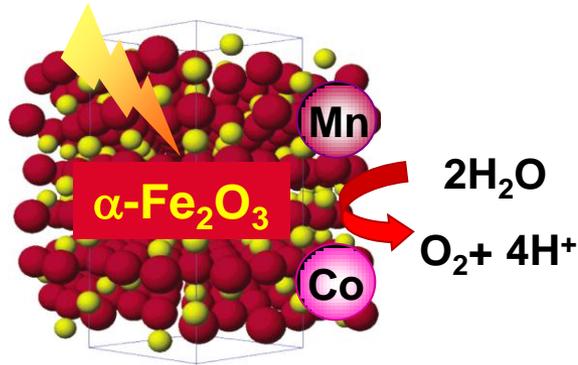
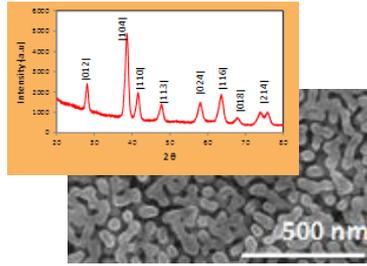
State of the art « PV + Electrolysis » technology

Sustainability
Economic viability



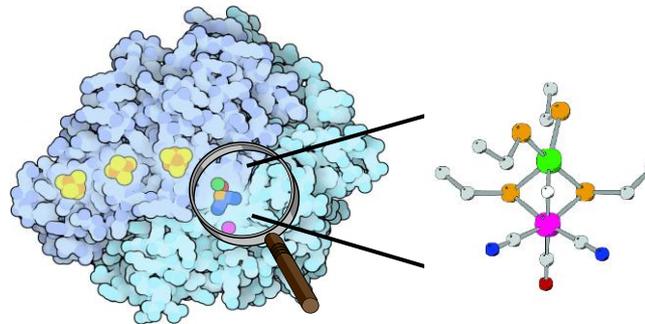
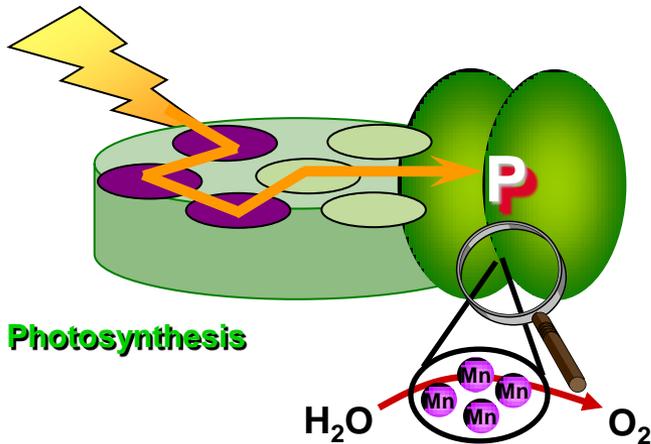
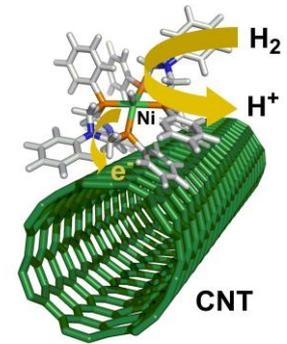
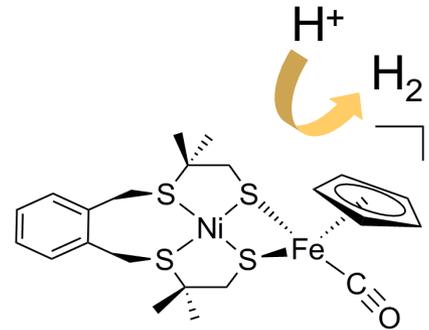
Fundamental research to TRL 2

Solar Fuels: a transversal field



Solid-state materials

Molecular chemistry Nanosciences



Biosciences

www.solar-fuels.org



The product:
The method:

a solar liquid fuel
being global and harness
the intellectual firepower of
government-funded research
institutions and R&D industry



Light capture

Water splitting



CO₂ catalysis

Driven academic research will lead to
simultaneous industry R&D and
commercially-viable products

The tools:

workshops
map of knowledge
fellow-exchange programs
translational prize (grants)



Advanced Materials and Processes for Energy Applications

43 participant/associate institutions

14 EU countries

This JP is structured as a matrix with 3 “tools” sub-programmes:

SP1 : Materials Science.

SP2 : Characterization of materials and processes.

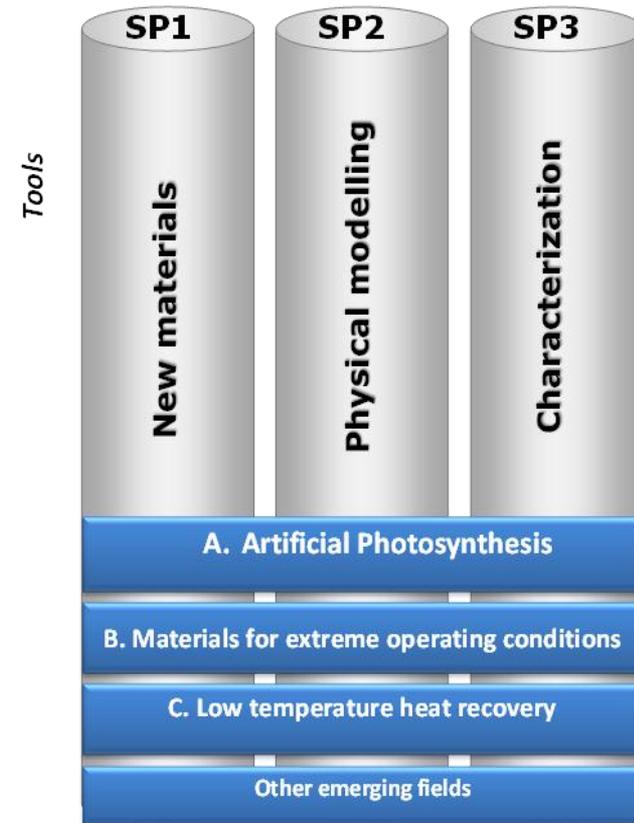
SP3 : Modelling of materials and processes

and “emerging applications,” among which

Artificial Photosynthesis appears as a first objective

Current activities in **Artificial Photosynthesis** might be grouped into three subfields:

- molecularly designed systems
- solid-state components
- nature-guided design



CM 1202 “Perspect-H₂O”: a COST Action on Supramolecular water splitting:

www.perspect-h2o.eu

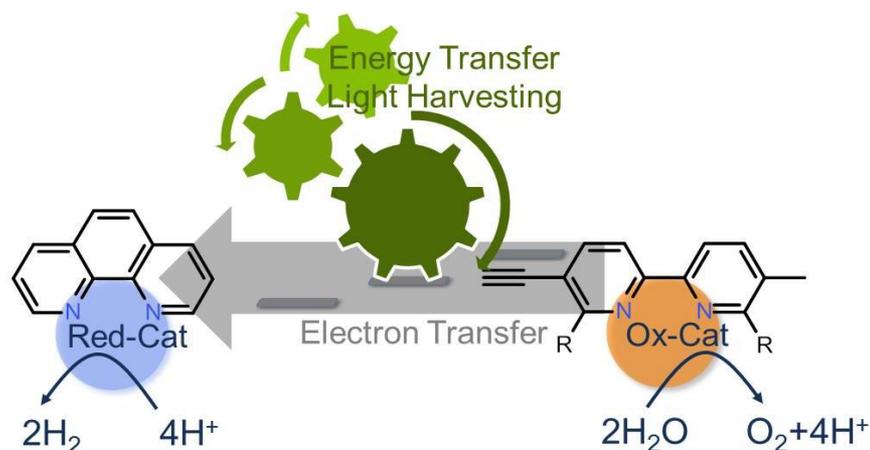
21 countries

55 research groups

Chair: Dr Benjamin Dietzek, IPHT-JENA



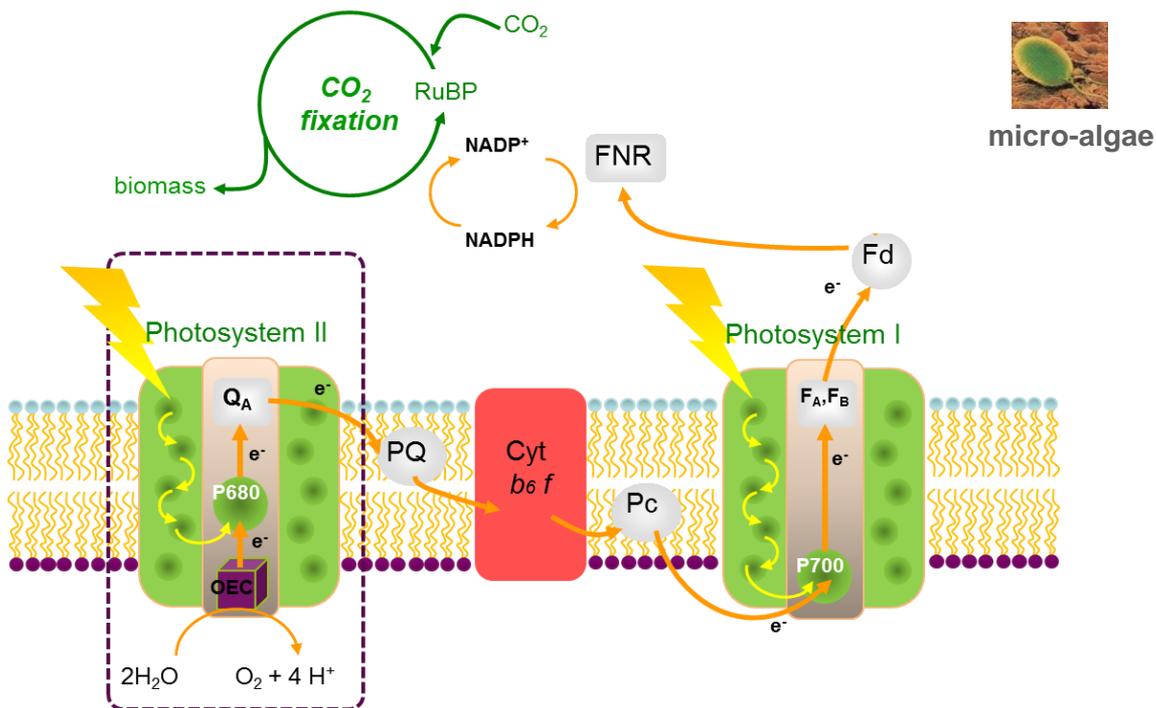
- fundamental understanding of the function determining light-induced elementary reactions in supramolecular photocatalytic water-splitting
- novel water-splitting supramolecular photocatalysts
- innovative functional systems operating in solutions, membranes or at surfaces



Working groups

- 1. Synthesis and Photocatalysis*
- 2. Device Integration*
- 3. Photoinduced Dynamics*
- 4. Intermediates & Active Species*

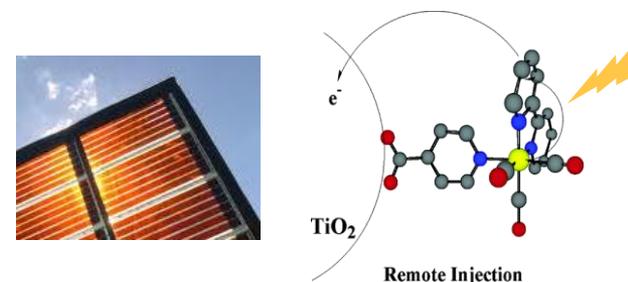
Photosynthesis



1. Conversion of light energy into electrochemical potential

Photon absorption and initial charge separation is achieved by molecular photosensitizers

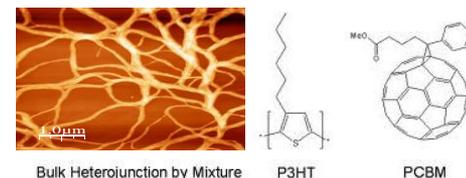
Spatial charge separation through a cascade of electron transfers



Dye-sensitized solar cells (Grätzel cells)

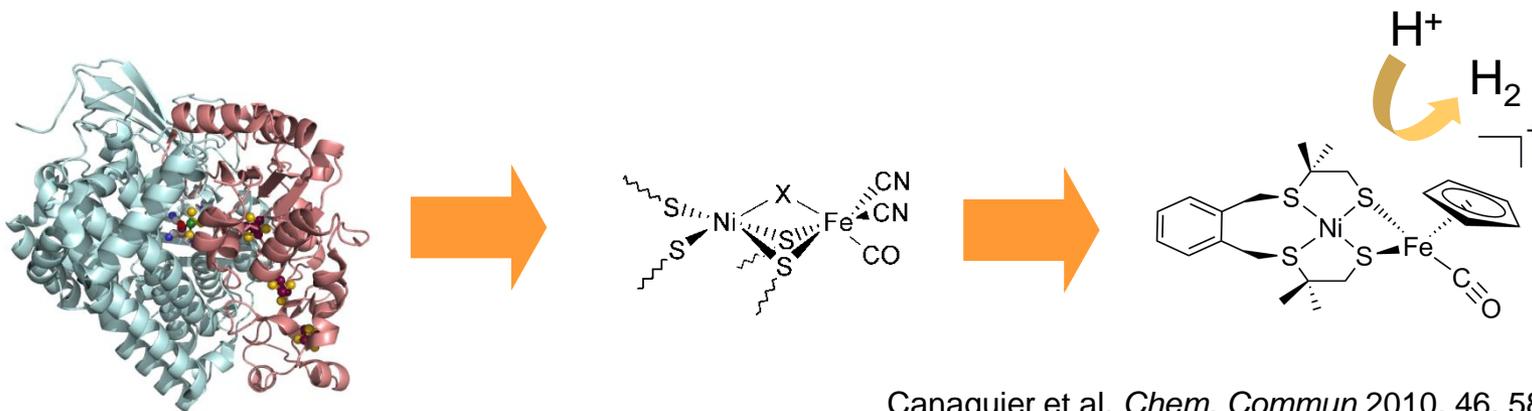
2. Efficient multielectron enzymatic catalysis

Fast
Close to the thermodynamic equilibrium (limited energetic loss)



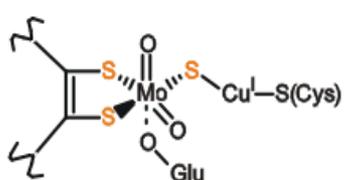
Organic photovoltaics

The biomimetic approach

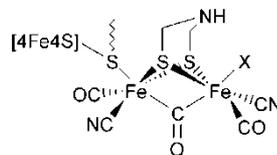
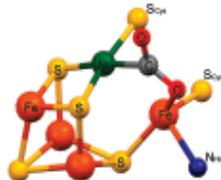


NiFe hydrogenases

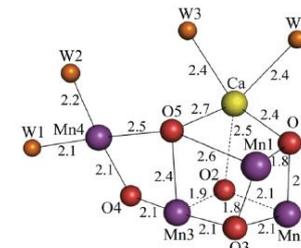
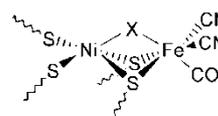
Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	Germanium
²⁴ Cr	²⁵ Mn	²⁶ Fe	²⁷ Co	²⁸ Ni	²⁹ Cu	³⁰ Zn	³¹ Ga	³² Ge
51.9961	54.938049	55.845	58.933200	58.6934	63.546	65.39	69.723	72.61
0.000044%	0.000000%	0.00294%	7.3×10 ⁻⁶ %	0.000161%	1.70×10 ⁻⁶ %	4.11×10 ⁻⁶ %	1.23×10 ⁻⁷ %	3.9×10 ⁻⁷ %
Niobium	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin
⁴² Mo	⁴³ Tc	⁴⁴ Ru	⁴⁵ Rh	⁴⁶ Pd	⁴⁷ Ag	⁴⁸ Cd	⁴⁹ In	⁵⁰ Sn
95.94	98.906250	101.071888	102.90550	106.363156	107.8682	112.411	114.818	118.710
8.3×10 ⁻⁹ %	0.000000%	6.1×10 ⁻⁵ %	1.1×10 ⁻⁵ %	4.5×10 ⁻⁵ %	5.8×10 ⁻⁹ %	5.3×10 ⁻⁹ %	6.0×10 ⁻¹⁰ %	1.25×10 ⁻⁸ %
Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	
⁷⁴ W	⁷⁵ Re	⁷⁶ Os	⁷⁷ Ir	⁷⁸ Pt	⁷⁹ Au	⁸⁰ Hg	⁸¹ Tl	⁸² Pb
183.84	186.207	190.23	192.217	195.078	196.966569	200.59	204.3833	207.2
4.34×10 ⁻¹⁰ %	1.69×10 ⁻⁹ %	2.20×10 ⁻⁹ %	1.6×10 ⁻⁹ %	4.4×10 ⁻⁹ %	6.1×10 ⁻⁹ %	1.11×10 ⁻⁹ %	6.0×10 ⁻¹⁰ %	1.03×10 ⁻⁸ %



CODH and formate reductase

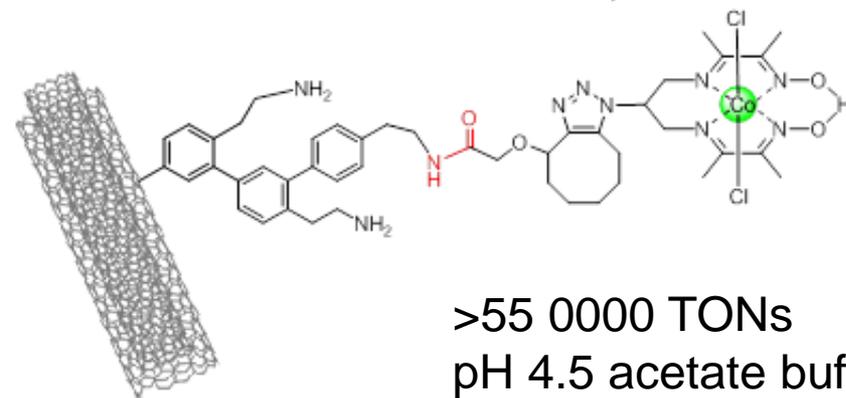
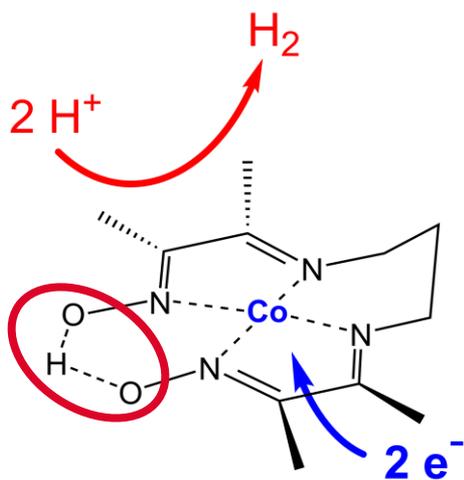
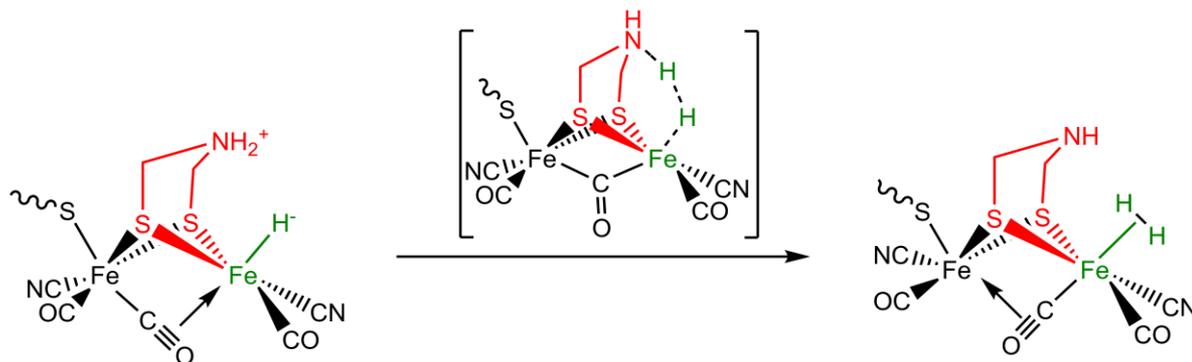


FeFe and NiFe hydrogenases



CaMn₄ Cluster (OEC)

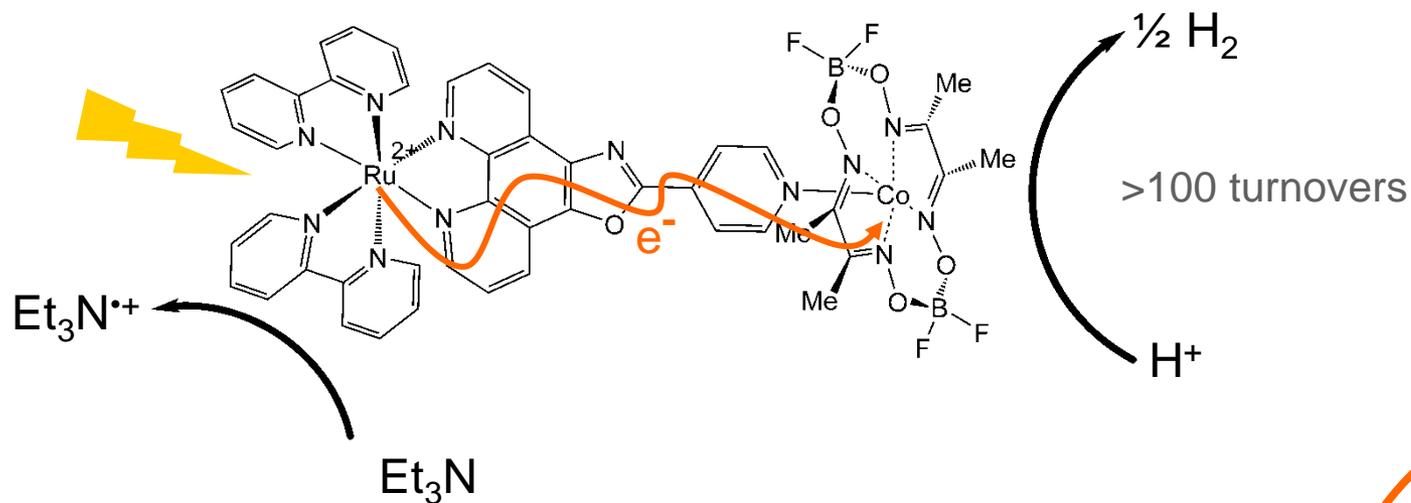
Bio-inspiration: Cobaloxime and cobalt dioxime-dioxime catalysts



>55 000 TONs
pH 4.5 acetate buffer
300 mV overpotential

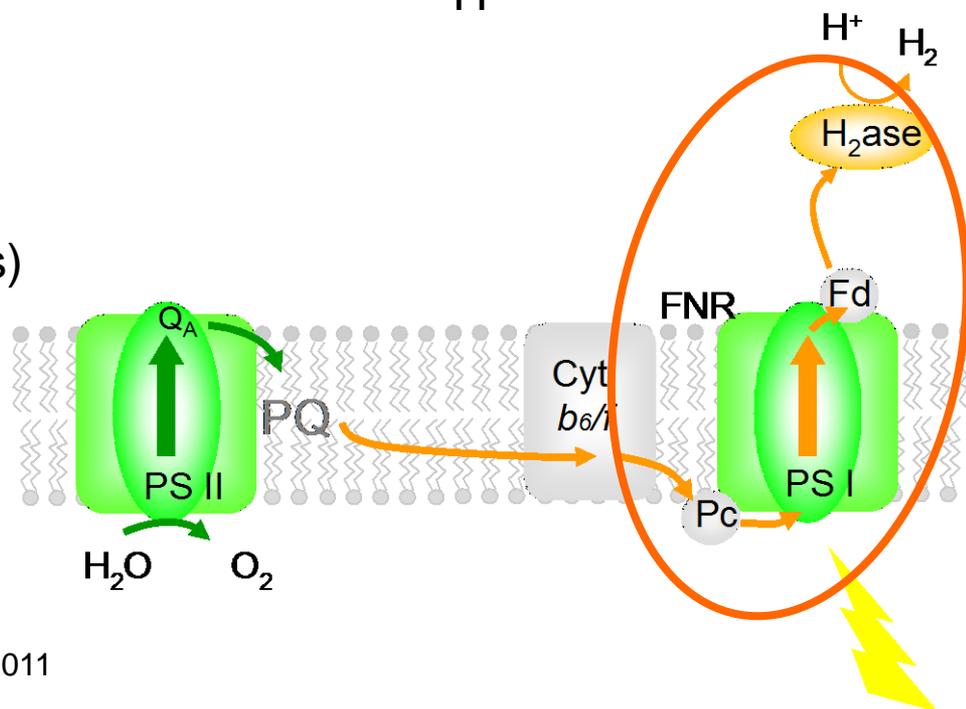
Molecular engineering of a cobalt-based electrocatalytic nanomaterial for H₂ evolution under fully aqueous conditions

Cobaloxime-based supramolecular photocatalysts



Single light-driven electron transfer (ps-ns)
Bi-electronic catalysis (μs-ms)

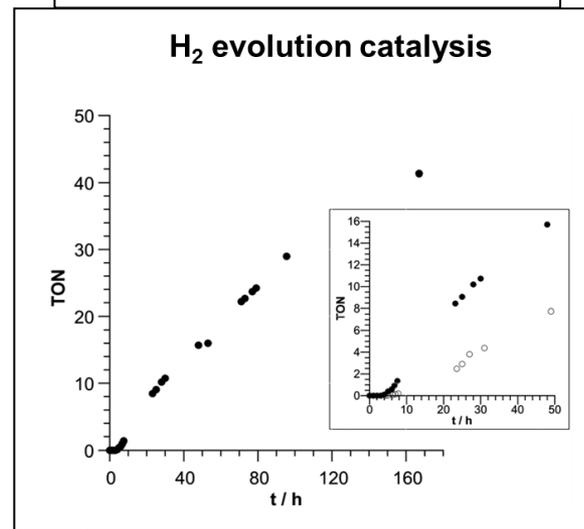
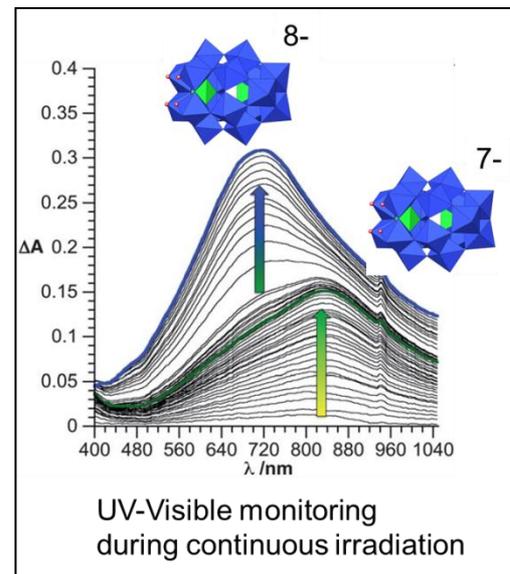
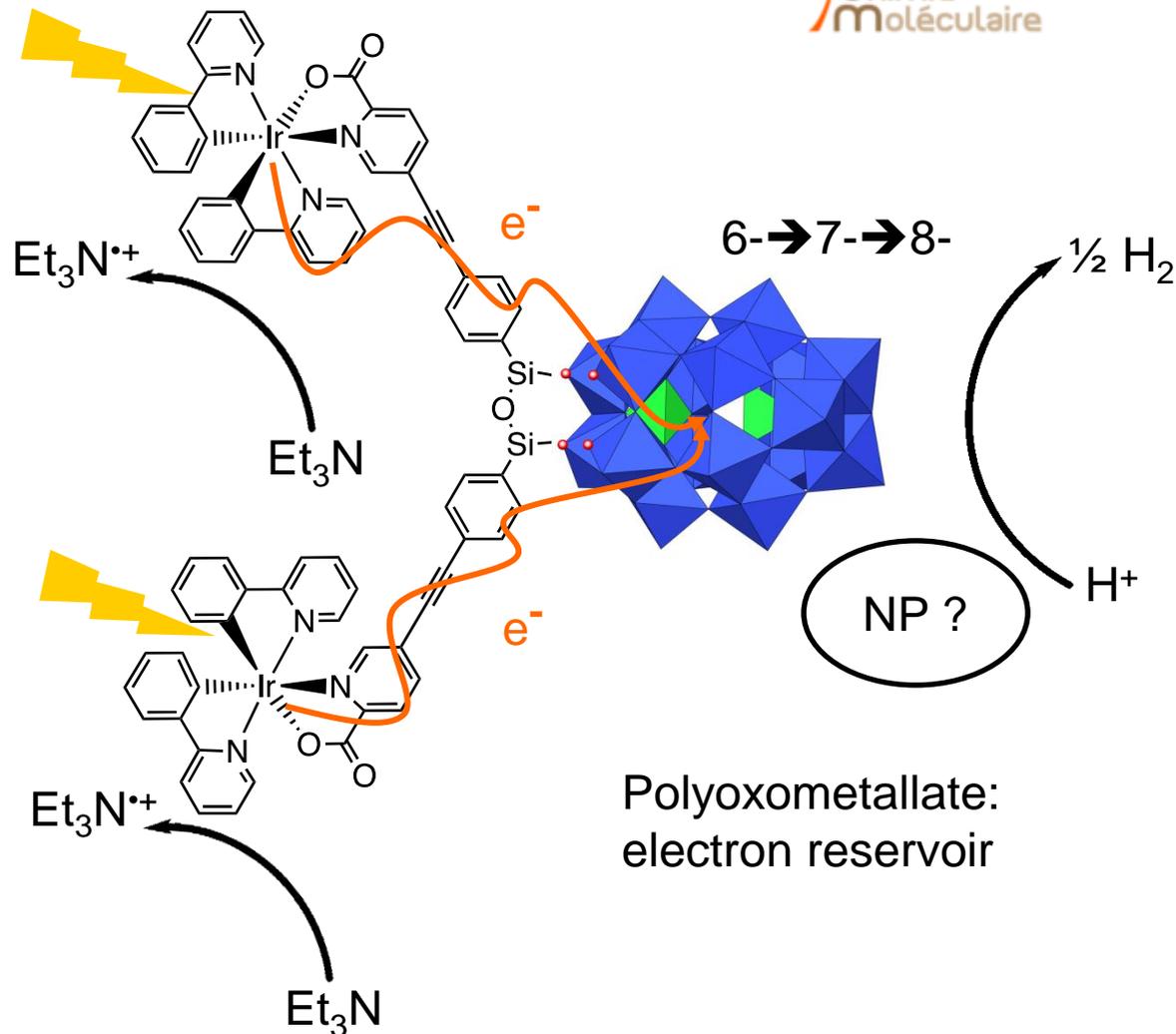
Fihri et al. *Angew. Chem. Int. Ed.* 2008



Review article

Artero, Chavarot-Kerlidou, Fontecave, *Angew. Chem. Int. Ed.* 2011

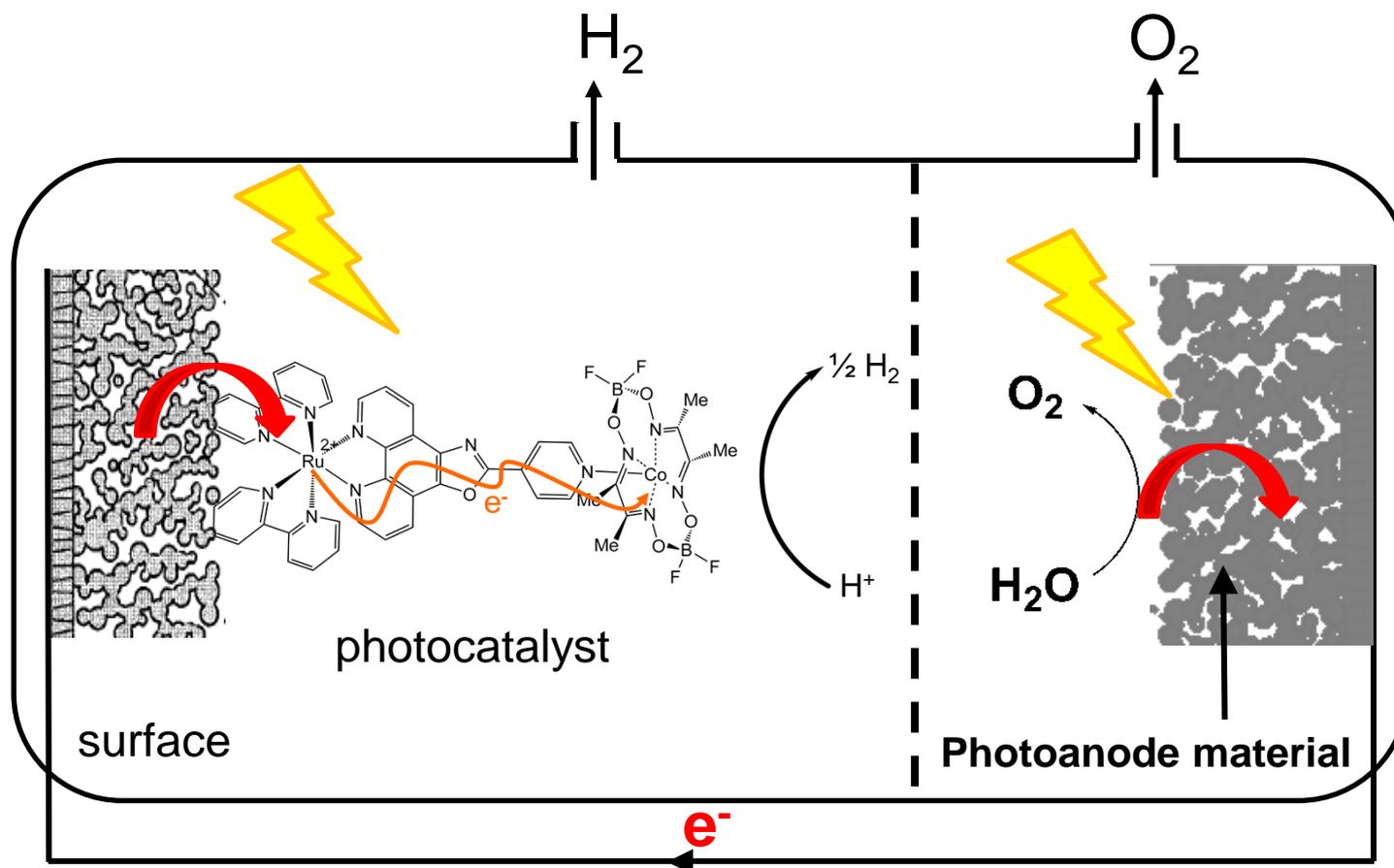
Charge photoaccumulation : towards improved systems



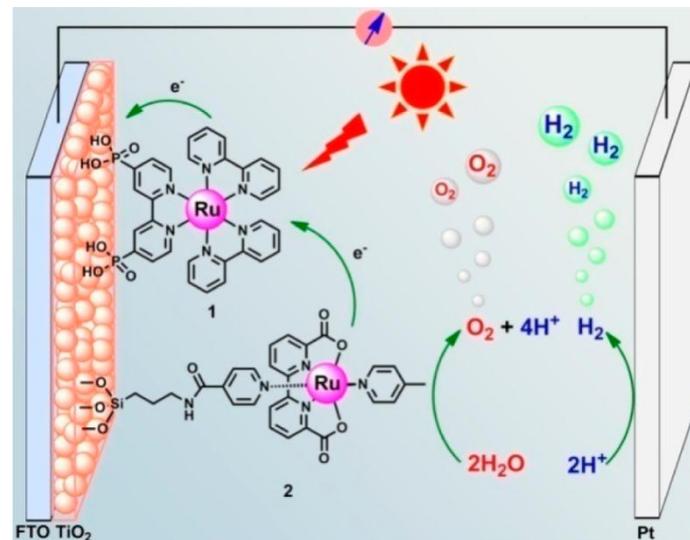
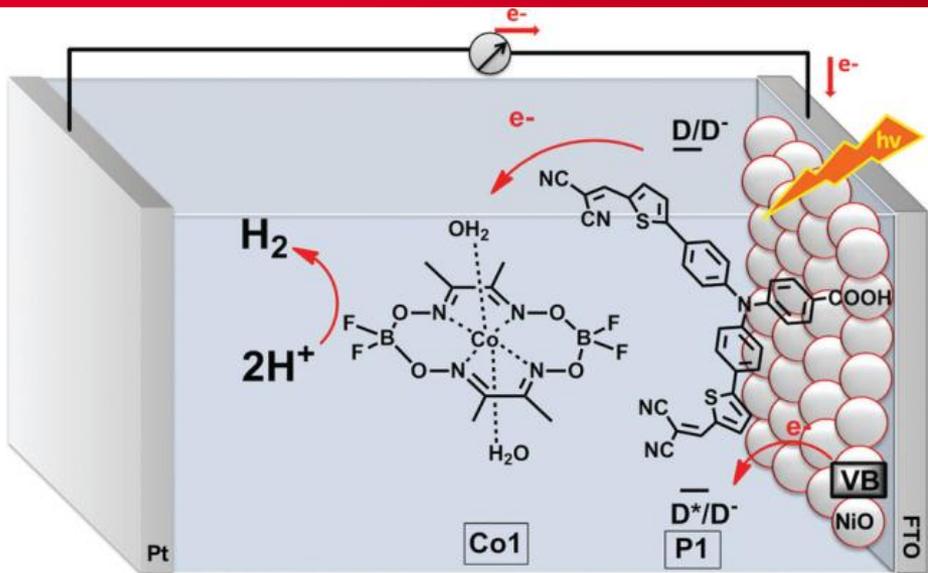
Towards Dye-Sensitized Photo-Electrosynthetic Cells (DSPEC)

PEC cells use renewable resources (water) to sustainably produce a fuel (H_2) and store renewable energy (sunlight)

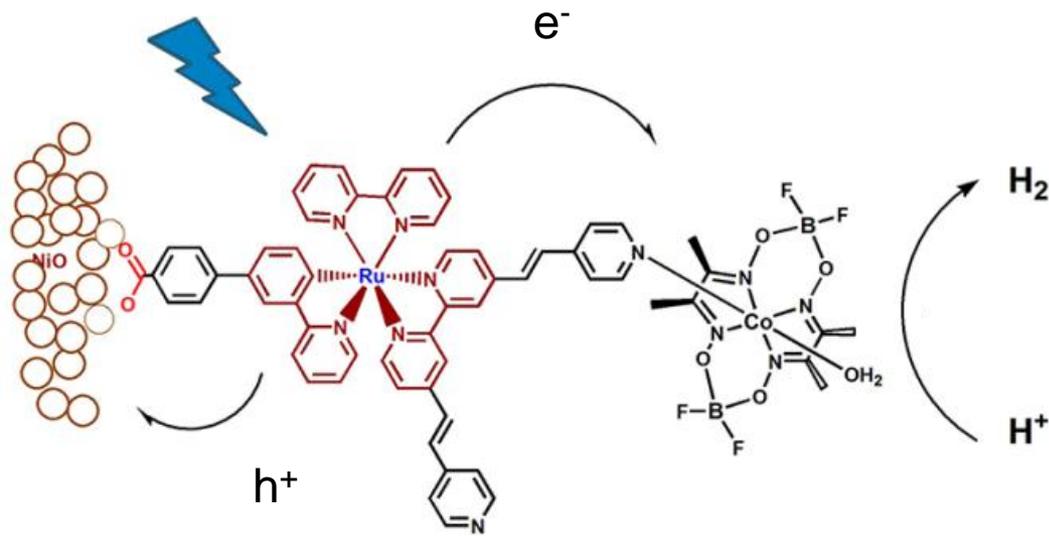
N. Lewis, Chem. Rev 2010
T. J. Meyer, Nature Chemistry 2011



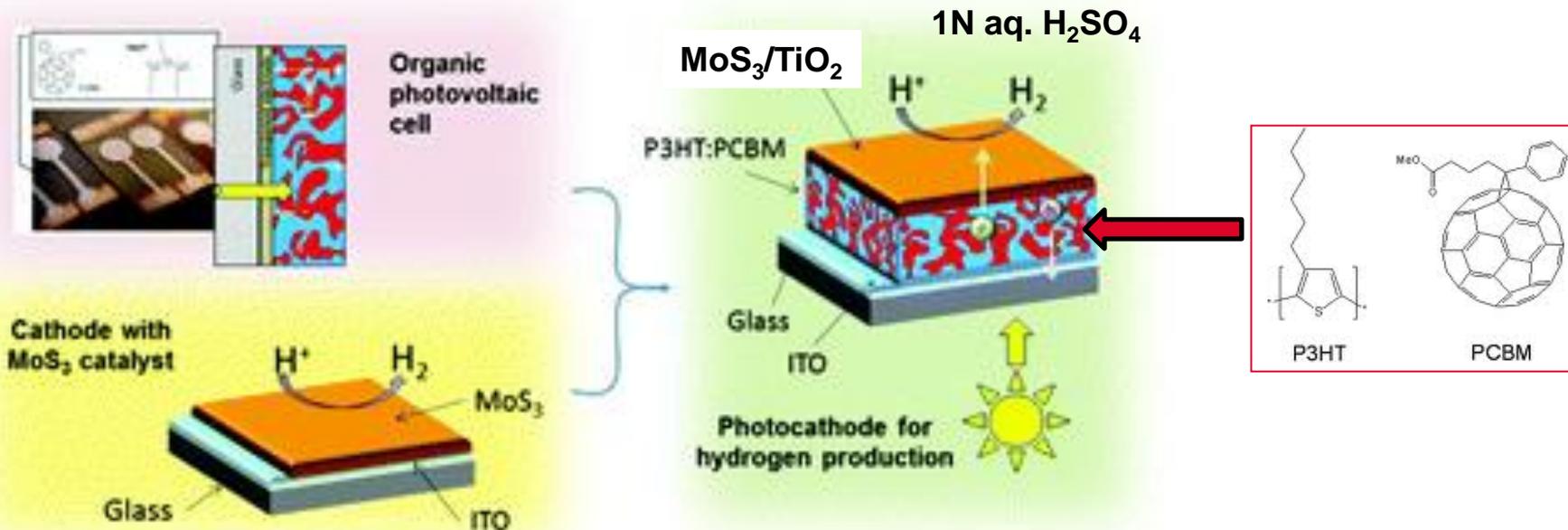
Towards Dye-Sensitized Photo-Electrosynthetic Cells (DSPEC)



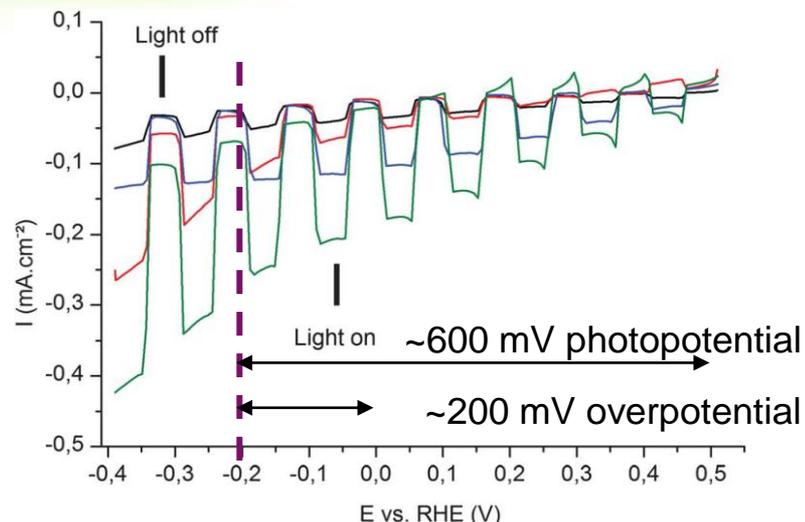
Photocathode
Yiying Wu and coll. *JACS* **2013**



Combining Organic Photovoltaics with water splitting catalysis



Collab. B. Jusselme, S. Palacin CEA-Saclay
Energy Environ. Sci. **2013**, 6, 2706



Conclusion

Progresses towards a mature hydrogen economy depends on breakthroughs in finding new catalytic materials

Biological systems provide the synthetic chemist with both challenges and inspiration regarding the design of new and efficient catalytic systems.

One of nature's most fundamental processes – photosynthesis – holds promises for turning the sun's energy into fuels.

Combining the bio-inspired approach with nanochemical tools, new and stable energy conversion materials based on earth-abundant elements can be derived.